

## Vsepr And Imf Homework

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### Vsepr And Imf Homework

VSEPR and IMF Homework Problems Lewis Structure Practice #1 Draw the Lewis structure for each of the following molecules: 1. Br 2 2. Methane 3. H<sub>2</sub>S 4. Nitrogen trichloride 5. Phosphorus pentabromide 6. Boron trihydride 7. Carbon disulfide Lewis Structure Practice #2 Draw the Lewis structure for each of the following molecules 1. Arsenic triiodide 2.

### VSEPR and IMF Homework

VSEPR: Electronic geometry of central atom(s)? Linear Trigonal Planar H: Linear F: Tetrahedral Trigonal Bipyramidal VSEPR: Molecular geometry of central atom(s)? Linear Linear Linear See-Saw Is the molecule polar? No No Yes Yes What is the predominate IMF? Dispersion

### IMF Unit - Lewis Structure, VSEPR Theory, VB Hybridization ...

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Lewis Structures, VSEPR, Polarity, IM Forces - Answers For each of the following molecules, draw the Lewis structure (with any resonance structures, if applicable), indicate the molecular shapes and bond angles, indicate the molecular polarity (if any), and identify the major intermolecular force in each compound. Hint - in this worksheet, as in

### Lewis Structures, VSEPR, Polarity, IM Forces

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### **Vsepr And Imf Homework**

HOMEWORK: 4/21 (B day) complete VSEPR review packet and crossword (see canal\_notes\_IMF) HOMEWORK: 4/22 (A day) complete unit 8 textwork and crossword; HOMEWORK: 4/25 (B day) complete study guide and study for test on 4/28; HOMEWORK: 4/26 (A day)

### **Unit10: Bonding and IMF - Mrs. Maureen Canal CP Physics ...**

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The molecular geometry of  $\text{NH}_3$  is trigonal pyramidal with asymmetric charge distribution on the central atom. Therefore this molecule is polar. Ammonia on Wikipedia. Back to Molecular Geometries & Polarity Tutorial: Molecular Geometry & Polarity Tutorial. For homework help in math, chemistry, and physics: [www.tutor-homework.com](http://www.tutor-homework.com).

### **Ammonia, $\text{NH}_3$ Molecular Geometry & Polarity**

4 Complete the table below. When drawing the VSEPR structures: Identify the most polar bond(s) by placine 8+ and 8-charges on their atoms • list bond angles VSEPR Drawings Line Drawings Bond Angles IMF(S) Present Methanol 1-Propanol 1-Butanol 2-Butanone Page 9 of 13

**Solved: 4 Complete The Table Below. When Drawing The VSEPR ...**

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