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There are four laws which govern the thermodynamic systems' phenomena, they are: Laws of Thermodynamics. First law of thermodynamics: When energy moves into or out of a system, the system's internal energy changes in accordance with the law of conservation of mass.; Second law of thermodynamics: The state of the entropy of the entire universe, as an isolated system, will always increase ...

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Solution: First law of thermodynamics,  $Q = \Delta U + W$ , gives  $\Delta U = Q - W$ . For an ideal gas, internal energy decreases due to decrease in temperature. Questions on First Law of Thermodynamics . Question 1: One gram of water is vaporized at 100 deg C and atmospheric pressure.

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