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Frank M. White is Professor Emeritus of Mechanical and Ocean Engineering at the University of Rhode Island. He studied at Georgia Tech and M.I.T. In 1966 he helped found, at URI, the first department of ocean engineering in the country. Known primarily as a teacher and writer, he has received eight teaching awards and has written four textbooks on fluid mechanics and heat transfer.

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A physical change does not change the chemical composition of matter. Melting, boiling, change of shape, or mass, and the formation of a mixture are examples of physical changes to matter. A chemical change changes the chemical composition of matter. Formation of new compounds from the reaction of other substances is an example. Chapter 1

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Hydrostatics - Wikipedia

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2.107 Mass of H₂SO₄ solution = 1.3028 g/mL × 40.00 mL = 52.112 g Total mass of Zn and H₂SO₄ solution before reaction = 9.520 g + 52.112 g = 61.632 g Total mass of solution after the reaction = 61.338 g Because of the conservation of mass, the difference between the two masses is the mass of H₂ produced.

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