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Calculate the energy released in joules when one mole of polonium-214 decays according to the equation [Atomic masses: Pb-210 = 209.98284 amu, Po-214 = 213.99519 amu, He-4 = 4.00260 amu.] A. 8.78 10¹⁴ J/mol

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The precipitate formed when barium nitrate and potassium chromate are placed in a test tube is... a) Barium Nitrate b) Barium Chromate c) Potassium Nitrate d) Potassium Chromate. ... 121.12 amu b) 149.12 amu c) 110.12 amu d) 243.03 amu. 149.09 amu. The empirical formula of C₁₀H₂₀ is: a) C₁₀H₂₀ b) C₅H₁₀ ... 6 answers. QUESTION. This principle ...

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Omit the milli and treat as gram amounts for easier math. Mass of burnt sample = 6.853 grams Mass of CO₂ produced = 20.08 grams Mass of H₂O produced = 5.023 grams Solution steps Step #1 Find the grams of carbon that produced the carbon dioxide Step #2 Find the grams of hydrogen that produced the water Step #3 Subtract the mass of carbon and ...

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What typical penetrator is used in Brinell hardness test? A. 10 mm ball B. 120° diamond (brale) C. 1 mm diameter ball D. 20°needle; What is the ratio of the maximum load in a tension test to the original cross-sectional area of the test bar? A. Tensile strength B. Yield strength C. Shear strength; D. Flexural Strength 94.

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